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Teacher Notes and
Answers Chapter 12

SECTION 1 SHORT

ANSWER 1. c 2. a 3. b

2. a. alcohol b. water c.
the gels 3. The mixture
is a colloid. The

properties are
consistent with those
reported in Table 3 on
page 404 of the text.

The particle size is
small, but not too
small, and the mixture

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Holt Modern Chemistry
Review CHAPTER 12 &
13: SOLUTIONS The following pages contain the bulk (but not all) of the information for the chapter 12 & 13 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.

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CHAPTER 12 REVIEW

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REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1.

Solid CaCl_2 does not conduct electricity.

Explain why it is considered to be an electrolyte. CaCl_2 is an ionic solid. In the crystal form, its ions are locked in position.

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Dissolving
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mc06se cFMsr i-vi

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author: kabdebumb

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chapter 12 section 2

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Chemistry Chapter 12

Review Solutions

Section 2 Answers.

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CHAPTER 1 REVIEW.

Matter and Change.

SECTION 1. SHORT

ANSWER Answer the

..... 12. 18. B. 18. 12. 6.

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Review multiple choice
(30) define & identify
suspensions &
solutions define an
alloy define & identify
electrolytes &
nonelectrolytes factors
that affect the rate of
dissolution definitions
of unsaturated,
saturated &
supersaturated
solutions general rules
for predicting whether
a solute is soluble in a

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solvent definition of
solubility effects of
temperature &
pressure on the
solubility of gases and
solids in liquids
definitions of molarity
(M) and molality (m)
solving ...

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Chapter 12-
Solutions

Chapter 12: Liquids
and Solids. STUDY.
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PLAY. Match. Gravity.

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Chemistry, G. Parham,
Masterman 2014-15
Modern Chemistry, Holt
et al. Terms in this set
(41) fluid. substance
that can flow and take
the shape of its
container, such as a
liquid or a gas ...
Chemistry Chapter 10:
States ...

**Chapter 12: Liquids
and Solids**

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Textbook**

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Modern Chemistry
Chapter 12-Solutions
Section 1- Types of
Mixtures Solutions are
homogeneous mixtures
of two or more
substances in a single
phase. Soluble
describes a substance
as capable of being
dissolved. Solvent is
the dissolving medium
in a solution. Solute is
the substance that is
dissolved in a solution.

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Change 2 1 2 3 ...
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Properties 6 Elements
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Scientific Method 27
Units of Measurement
31 Using Scientific

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378 1 2 3 Types of ...

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Particle Mass number

19. Proton 20. Neutron

21. Electron 10

CHAPTER 3 TEST

Relative charge

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Class CHAPTER 3 TEST
continued SHORT
ANSWER Write the
answers to the
following questions in
the space provided. 22.

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Practice Test**

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164 Study Guide for An
Introduction to
Chemistry Section
Goals and
Introductions Section
11.1 The Mysterious
Electron Goals To
explain why it is very
difficult to describe the
modern view of the
electron. To give you
some understanding of
the nature of the
electron by describing
how it is like a guitar

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string. To explain what
atomic orbitals are.

1

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